Chemical and Physical Properties and Changes

Warm - Up

What is the difference between qualitative and quantitave? Give 3 examples of each one. Previously...

- We talked about matter and physical properties.
- Qualitative vs Quantitative

The other property of matter is a chemical property.

Chemical property - any property that describes how a substance reacts with another substance when forming something new.

Examples of a chemical property:

- Combustability
- Reacting with acids
- Reacting with air
- Reacting with water

Chemical vs Physical Change

Look at pages 28-29 in your book and write down the difference between these two types of changes.

With a physical change, the substance remains the same. With a chemical change. We end up having a new substance with different properties.

How do you know a chemical change has happened?

- 1. A new color appears
- 2. Heat or light is given off
- 3. Bubbles of gas are formed
- 4. A precipitate (solid) forms in a liquid.
- 5. The change is difficult to reverse

Classify each of these as a chemical or physical change. Explain your answers.

(a) Rust (b) Igniting a campfire(c) Shredding paper(d) Bleaching clothes

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Physical versus Chemical Change handout