

Copy and complete the following table

atom/ion symbol	atomic number (Z) $Z = p$	mass number (A) $A = p + n$	protons $p (+)$	electrons $e(-)$	neutrons n	charge (compare $p + e$)
	9	19				0
			20	18	21	
Fe^{3+}		56				3+
	35			36	45	
C		12				0
		14		2	8	
Zn^{2+}		64				2+
	30			30	42	