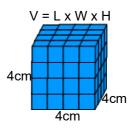
**Particle Theory** 



1. The density of the block is  $85 \text{ g/cm}^3$ . The volume can be found by looking at the block to the right. Calculate the mass.



2. The weight of water in a glass in 70 grams. The volume of the water is 70 cm<sup>3</sup>. What is the density?

## Particle Theory - Pg 44 - 47

## Answer the questions below using complete sentences

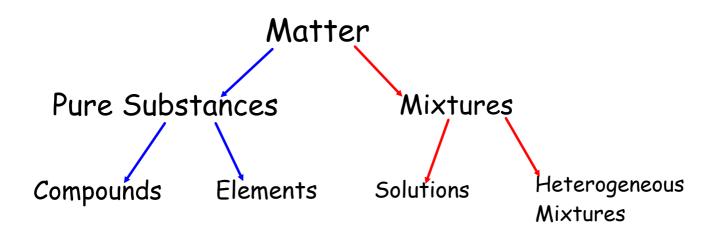
1. What are the four principles of particle theory. Draw a picture to illustrate each one. Also, write each rule.

2. What is the difference between a pure substance and a mixture? Give 2 examples of each.

3. Mixtures can be broken down into two sub-categories. What are they and how are they different from each other?

4. Pure substances can be broken down into two subcategories. What are they and how are they different from each other?

5. Make a chart that includes the following terms: Matter, pure substance, mixture, solution, heterogeneous mixture, element and compund



Elements - Can<u>not</u> be broken down (only one kind of atom)

Compounds - <u>Can</u> be broken down (two or more types of atoms)

NOTE: A pure substance can ONLY be seperated chemically **Particle Theory** 

## Solutions - homogeneous mixture "seems as one" It only has one visible phase. (Ex.

## Heterogeneous Mixture - You

see two or more phases or

visible parts. (Ex.

