

Write the name or chemical formula for each of the following.

1.  $\text{Ca}(\text{OH})_2$  \_\_\_\_\_
2.  $\text{K}_2\text{SO}_4$  \_\_\_\_\_
3. potassium nitrite \_\_\_\_\_
4.  $\text{MgCO}_3$  \_\_\_\_\_
5. potassium carbonate \_\_\_\_\_
6.  $\text{Ag}_2\text{SO}_4$  \_\_\_\_\_
7. ammonium dichromate \_\_\_\_\_
8. iron(III) phosphate \_\_\_\_\_
9.  $\text{NH}_4\text{OH}$  \_\_\_\_\_
10.  $\text{Ni}(\text{OH})_2$  \_\_\_\_\_
11.  $\text{Fe}(\text{NO}_3)_3$  \_\_\_\_\_
12.  $\text{CoCO}_3$  \_\_\_\_\_
13. potassium nitrate \_\_\_\_\_
14. potassium nitrite \_\_\_\_\_
15.  $\text{Al}(\text{OH})_3$  \_\_\_\_\_
16.  $\text{Ni}(\text{NO}_3)_2$  \_\_\_\_\_
17.  $\text{NaOH}$  \_\_\_\_\_
18. sodium nitrite \_\_\_\_\_
19. potassium sulfate \_\_\_\_\_
20. iron(II) hydroxide \_\_\_\_\_
21.  $\text{Cu}(\text{CH}_3\text{COO})_2$  \_\_\_\_\_
22. potassium phosphate \_\_\_\_\_
23. iron(II) sulfite \_\_\_\_\_
24. sodium sulfate \_\_\_\_\_
25. magnesium nitrate \_\_\_\_\_
26. iron (II) carbonate \_\_\_\_\_
27. potassium hydroxide \_\_\_\_\_
28. magnesium sulfate \_\_\_\_\_
29. iron(II) sulfate \_\_\_\_\_
30.  $\text{Cr}(\text{NO}_3)_3$  \_\_\_\_\_
31. strontium nitrate \_\_\_\_\_
32.  $\text{Sr}(\text{OH})_2$  \_\_\_\_\_
33. ammonium nitrate \_\_\_\_\_
34. ammonium sulfide \_\_\_\_\_
35. mercury (II) carbonate \_\_\_\_\_
36. tin (IV) phosphate \_\_\_\_\_
37. copper (II) carbonate \_\_\_\_\_
38. iron (III) sulfate \_\_\_\_\_
39. zinc nitrate \_\_\_\_\_
40. ammonium sulfate \_\_\_\_\_
41. silver permanganate \_\_\_\_\_
42. iron (I) nitrate \_\_\_\_\_
43. sodium acetate \_\_\_\_\_
44. aluminum nitrate \_\_\_\_\_
45. aluminum sulfate \_\_\_\_\_
46. ammonium dihydrogen phosphate \_\_\_\_\_